

corrections were taken from *International Tables for X-ray Crystallography* (1974, Vol. IV, pp. 71, 148). All computer programs were from the *TEXSAN* package (Molecular Structure Corporation, 1985). Table 1 lists the final atomic coordinates and equivalent isotropic temperature factors; Table 2 presents selected interatomic distances and angles.* Fig. 1 shows the atom-numbering scheme and the thermal motion of the molecule; Fig. 2 illustrates the molecular packing.

Related literature. Equilibrium acidity data [pK_a values in dimethyl sulfoxide (DMSO) solution] for five variously *G*-substituted aryl-1,3-propanedinitriles (where *G* = 3-CN, 4-Cl, 4-H, 4-CH₃, 4-OCH₃) as well as the title compound (*G* = 4-NH₂) are 1.9,

* Lists of structure factors, anisotropic thermal parameters, intermolecular distances, torsion angles and selected least-squares planes have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54580 (9 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: CR0366]

3.14, 4.24, 4.85, 5.68 (Arnett, Chawla, Amarnath & Whitesell, 1987) and 7.24 (Jirka, 1989), respectively. These acidity data plot linearly with Hammett σ . It is therefore likely that the title compound is deprotonated at C(2), rather than at N(4), in DMSO solution. The X-ray structure for the title compound indicates that an H atom is bound at C(2).

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N-(4,6-Diméthylpyridin-2-yl)benzèthiocarboxamide

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Abstract. C₁₄H₁₄N₂S, $M_r = 242.3$, monoclinic, $C2/c$, $a = 20.004$ (2), $b = 7.997$ (1), $c = 15.954$ (1) Å, $\beta = 90.40$ (1)°, $V = 2552.2$ (7) Å³, $Z = 8$, $D_x = 1.261$ Mg m⁻³, $\lambda(\text{Cu } K\alpha) = 1.5418$ Å, $\mu = 2.0$ mm⁻¹, $F(000) = 1024$, $T = 295$ (1) K, $R = 0.039$ for 1807 independent observed reflections. There is a delocalized orbital over the N=C=S thioamide group. The pyridinyl and the phenyl rings are approximately planar and the dihedral angles between their least-squares planes and the NCS thioamide plane are 25.2 (2) and 44.2 (2)° respectively. The molecules form pairs in which they are linked to each other by two weak N(7)—H(7)···N(1') hydrogen bonds [3.192 (3) Å, 177 (3)°] ($i: -x, y,$

$\frac{3}{2} - z$). The title compound shows only a very moderate anti-inflammatory activity, even though it is the thio analog of *N*-(4,6-dimethylpyridin-2-yl)benzamide which is a non-acidic compound exhibiting a potent anti-inflammatory activity.

Partie expérimentale. Cristal parallélépipédique: 0,13 × 0,22 × 0,26 mm. Dimensions de la maille déterminées avec 25 réflexions pour lesquelles 17,55 ≤ θ ≤ 32,40°. Diffractomètre Enraf-Nonius CAD-4, balayage $\theta/2\theta$, 0,023 ≤ $(\sin\theta)/\lambda$ ≤ 0,573 Å⁻¹; 0 ≤ h ≤ 22, 0 ≤ k ≤ 9, -18 ≤ l ≤ 17. Réflexions de contrôle de l'intensité: 10, 2, 1, 117 et 517. Variations non significatives de I au cours des mesures. 2007 réflex-

Tableau 1. *Coordonnées atomiques relatives, facteurs de température isotropes équivalents et écarts-type*

$$B_{\text{eq}} = (4/3)[\beta_{11}a^2 + \beta_{22}b^2 + \beta_{33}c^3 + \beta_{12}ab(\cos\gamma) + \beta_{13}ac(\cos\beta) + \beta_{23}bc(\cos\alpha)].$$

	x	y	z	$B_{\text{eq}} (\text{\AA}^2)$
N(1)	-0,03001 (9)	0,3848 (2)	0,6449 (1)	3,32 (4)
C(2)	0,0184 (1)	0,2898 (3)	0,6126 (1)	3,03 (4)
C(3)	0,0179 (1)	0,2288 (3)	0,5309 (1)	3,41 (5)
C(4)	-0,0352 (1)	0,2751 (3)	0,4791 (1)	3,65 (5)
C(5)	-0,0834 (1)	0,3797 (3)	0,5105 (2)	3,93 (5)
C(6)	-0,0803 (1)	0,4314 (3)	0,5937 (2)	3,70 (5)
N(7)	0,06890 (8)	0,2550 (3)	0,6726 (1)	3,16 (4)
C(8)	0,1329 (1)	0,2059 (3)	0,6634 (1)	2,94 (4)
C(9)	0,1668 (1)	0,1694 (3)	0,7447 (1)	3,01 (4)
C(10)	0,1338 (1)	0,0773 (3)	0,8059 (1)	3,68 (5)
C(11)	0,1653 (1)	0,0436 (4)	0,8817 (2)	4,84 (6)
C(12)	0,2283 (1)	0,1024 (4)	0,8975 (2)	5,47 (7)
C(13)	0,2615 (1)	0,1931 (5)	0,8371 (2)	5,74 (7)
C(14)	0,2313 (1)	0,2248 (4)	0,7599 (2)	4,32 (6)
S(15)	0,17410 (3)	0,1851 (1)	0,57479 (4)	4,61 (1)
C(16)	-0,0377 (1)	0,2132 (4)	0,3894 (2)	5,14 (6)
C(17)	-0,1320 (1)	0,5440 (4)	0,6309 (2)	5,32 (6)

Tableau 2. *Longueurs (Å) et angles des liaisons (°) et écarts-type*

N(1)—C(2)	1,337 (3)	N(7)—C(8)	1,348 (3)
N(1)—C(6)	1,344 (3)	C(8)—C(9)	1,487 (3)
C(2)—C(3)	1,391 (3)	C(8)—S(15)	1,650 (2)
C(2)—N(7)	1,415 (3)	C(9)—C(10)	1,393 (3)
C(3)—C(4)	1,391 (3)	C(9)—C(14)	1,384 (3)
C(4)—C(5)	1,374 (3)	C(10)—C(11)	1,386 (3)
C(4)—C(16)	1,515 (4)	C(11)—C(12)	1,368 (4)
C(5)—C(6)	1,392 (3)	C(12)—C(13)	1,379 (4)
C(6)—C(17)	1,498 (4)	C(13)—C(14)	1,392 (4)
C(2)—N(1)—C(6)	117,6 (2)	C(2)—N(7)—C(8)	131,2 (2)
N(1)—C(2)—C(3)	124,0 (2)	N(7)—C(8)—C(9)	113,0 (2)
N(1)—C(2)—N(7)	111,5 (2)	N(7)—C(8)—S(15)	127,0 (2)
C(3)—C(2)—N(7)	124,4 (2)	C(9)—C(8)—S(15)	120,0 (2)
C(2)—C(3)—C(4)	117,6 (2)	C(8)—C(9)—C(10)	120,0 (2)
C(3)—C(4)—C(5)	118,7 (2)	C(8)—C(9)—C(14)	120,6 (2)
C(3)—C(4)—C(16)	119,7 (2)	C(10)—C(9)—C(14)	119,4 (2)
C(5)—C(4)—C(16)	121,6 (2)	C(9)—C(10)—C(11)	120,0 (2)
C(4)—C(5)—C(6)	120,1 (2)	C(10)—C(11)—C(12)	120,4 (2)
N(1)—C(6)—C(5)	121,8 (2)	C(11)—C(12)—C(13)	120,0 (2)
N(1)—C(6)—C(17)	116,2 (2)	C(12)—C(13)—C(14)	120,4 (3)
C(5)—C(6)—C(17)	122,0 (2)	C(9)—C(14)—C(13)	119,8 (2)

ions indépendantes mesurées, 200 réflexions inobservées [$I < 3\sigma(I)$]. Programme *MULTAN11/82* (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982). Affinement basé sur les F . Programme à matrice complète. Facteurs de diffusion des *International Tables for X-ray Crystallography* (1974, Tome IV, pp. 99, 149). Paramètres affinés: x, y, z de tous les atomes et β_{ij} de C, N et S. B de chaque H pris égal à (B_{eq} de l'atome lié à H) $\times 1,3$. $R = 0,039$, $wR = 0,038$, $w = 1$ pour toutes les réflexions, $S = 1,31$, g (coefficient d'extinction secondaire isotrope) = $1,33 (7) \times 10^{-6}$, $(\Delta/\sigma)_{\text{max}} = 0,05$, $\Delta\rho_{\text{max}} = 0,26 (4)$, $\Delta\rho_{\text{min}} = -0,32 (4)$ e \AA^{-3} . Les valeurs maximale et minimale de la densité électronique résiduelle s'observent au voisinage de l'atome de soufre. Programmes de calcul du système *SDP* (B. A. Frenz

& Associates, Inc. 1982). Figs. 1 et 2: *ORTEPII* (Johnson, 1976).

Les coordonnées atomiques relatives et les facteurs de température isotrope équivalents sont rapportés dans le Tableau 1, les principales distances interatomiques et les angles des liaisons dans le Tableau 2.* La Fig. 1 représente la molécule vue en perspective et la Fig. 2 la projection de la structure sur la face (010). Les noms des atomes sont indiqués sur la Fig. 1.

* Les listes des facteurs de structure, des coefficients d'agitation thermique anisotrope des 'atomes lourds', des coordonnées des atomes d'hydrogène, des distances C—H et N—H, des angles de torsion, des distances des atomes aux plans moyens et des distances interatomiques intermoléculaires ont été déposées aux archives de la British Library Document Supply Centre (Supplementary Publication No. SUP 54527 (14 pp.)). On peut en obtenir des copies en s'adressant à: The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, Angleterre. [Référence de CIF: PA0251]

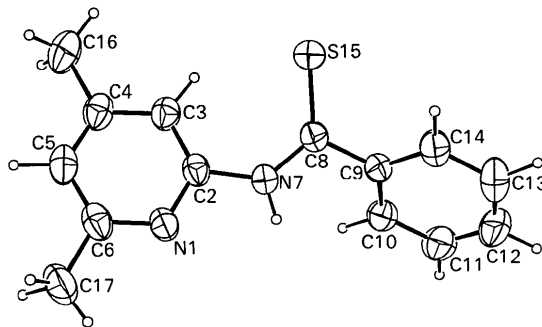


Fig. 1. Dessin de la molécule vue en perspective et noms des atomes.

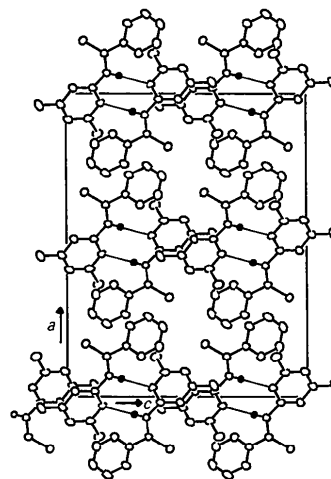


Fig. 2. Dessin de la structure vue selon [010]. Les cercles noirs représentent les atomes H(7) et les traits les plus fins les liaisons hydrogène N(7)—H(7)···N(1').

Littérature associée. Structure cristalline du *N*-(4,6-diméthylpyridin-2-yl)benzamide (Rodier, Piessard, Le Baut & Michelet, 1986). Structure cristalline du *N*-éthyl-*N*-(4,6-pyridin-2-yl)benzamide (Rodier, Piessard, Le Baut & Brion, 1987).

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Structure of *N,N*-Bis(2,3,4,5,6-Pentafluorophenyl)oxamide

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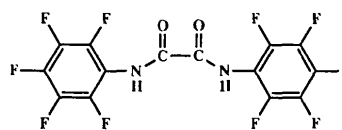
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Abstract. C₁₄H₂F₁₀N₂O₂, *M_r* = 420.16, monoclinic, *P*2₁/*n*, *a* = 14.206 (1), *b* = 4.990 (1), *c* = 10.964 (3) Å, β = 111.68 (1)°, *V* = 722.3 (1) Å³, *Z* = 2, *D_x* = 1.932 Mg m⁻³, λ(Cu Kα₁) = 1.54050 Å, μ = 1.974 mm⁻¹, *F*(000) = 412, *T* = 295 K, final *R* = 0.040 for 971 reflections. The midpoint of the oxamide carbon chain lies on a crystallographic inversion center. An intermolecular hydrogen bond, N—H⋯O [N(1)—O(3) 2.877 (2) Å], was observed.

Experimental. (I) was obtained by the reaction of pentafluoroaniline with oxalyl chloride in C₆H₆. Recrystallization from CH₃(CH₂)₄CH₃/CH₃COO-C₂H₅ gave colorless prisms (0.30 × 0.30 × 0.45 mm) having m.p. 515.5–517.0 K. The combustion analysis of this compound was consistent with the structure. Rigaku AFC-5 four-circle diffractometer used with ω-2θ-scan method, ω-scan width (1.3 + 0.14 tan θ)° and scan speed 16° min⁻¹. Lattice parameters obtained from least-squares analysis of 20 reflections with 2θ values ranging from 55 to 61°. Of 1323 reflections scanned within index range *h* -15–15, *k* 0–5, *l* 0–12 up to sin θ/λ < 0.56 Å⁻¹, *R*_{int} = 0.018 for 69 reflections, 1072 unique reflections [*F* > σ(*F*)] classified as observed. Three standard reflections measured every 150 reflections, intensity variation < 3%. Intensities corrected for Lorentz and polarization factors, but absorption correction not applied. Structure solved using program package *SAP*85

(Yao, Zheng, Qian, Han, Gu & Fan, 1985), a version of *MULTAN*80 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980). The refinement was carried out by the full-matrix least-squares method with anisotropic temperature factors for non-H atoms. The function minimized was $\sum w[(|F_o|)^2 - (|F_c|)^2]^2$ with $w = 1/[\sigma^2(F_o) + 0.016(F_o)]$; σ(*F_o*) was determined from counting statistics. All H atoms located from a difference map and theoretical calculations were refined, initial thermal parameters set at equivalent isotropic thermal parameters of each bonded atom. Final discrepancy indices *R* = 0.040, *wR* = 0.042, *S* = 2.018 for 136 variables and 971 reflections with *F* > 3σ(*F*). Maximum Δ/σ = 0.27 in final least-squares cycle. Final difference Fourier excursions 0.19 and -0.23 e Å⁻³. All major computations performed on a PANAFACOM computer with the *RCRYSTAN* (Rigaku Corporation, 1985) X-ray analysis program system. The atomic scattering factors were taken from *International Tables for X-ray Crystallography* (1974, Vol. IV).



(I)